What Is Mercury & Why Should We Remove It From Our Stacks?

- Liquid Metal
- 113 Pounds/gallon
- Neurotoxin (The Mad Hatter)
- Dimethylmercury Cancer in mice



Where Does Hg Originate?

- Cinnabar, HgS
- Largest emission is from coal fired power plants
- $HgS + O_2 >> SO_{2(g)} + Hg_{(g)}$
- O2 + Hg(g) >> + HgO(g)
- From Sludge



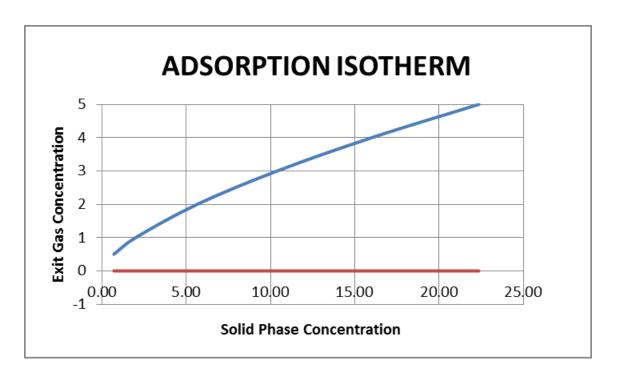
New Federal Limits

Effective March 21, 2016

NEW Hg EMISSIONS LIMITS		
	Exhisting	New
Fluid Beds	0.037	0.0010
Multiple Hearths	0.28	0.0010
Units	mg/dscm @ 7% O ₂	

How Do We Capture Hg?

- A<u>d</u>sorption (film on your windshield)
- Similar to Activated Carbon (AC) filled odor scrubbers. But with chemical reaction.



Unit Operations

- 1. Remove the heat
- 2. Remove the large particulate
- 3. Remove water droplets
- 4. Heat to >20°F above the dew point
- 5. Remove ultra-fine particles
- 6. Adsorb the Hg on Activated Carbon



1. Remove The Heat

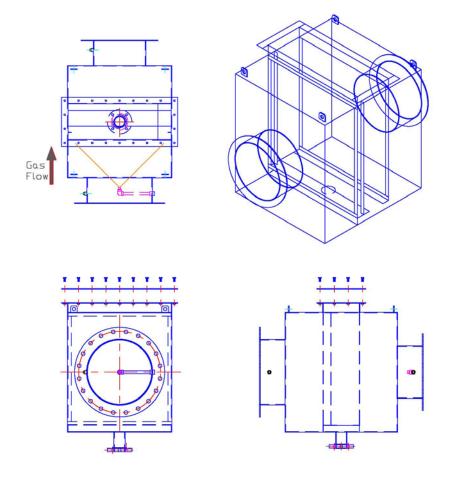
- Here in NJ every location uses a wet scrubber.
 - Venturi scrubber
 - Tray scrubber
 - VenturiPAK scrubber
 - Ring Jet scrubber
 - Packed bed scrubber
- The result is a saturated gas at 80°F 100°F.

2. Remove The Large Particulate

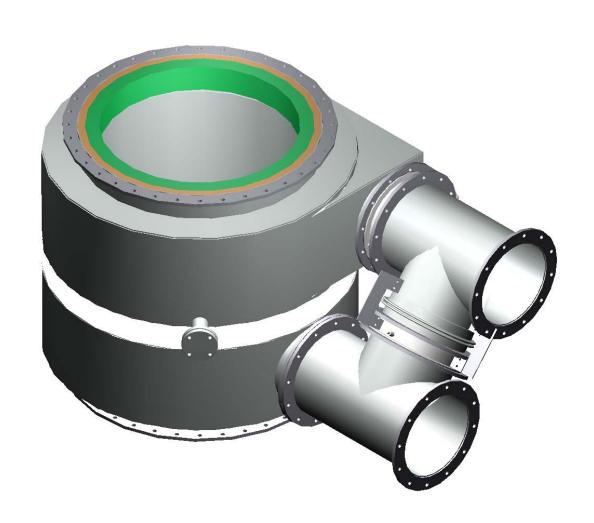
- This is most often done in the wet scrubber system.
- Frequently, a Wet Electrostatic Precipitator (WESP) is added following the scrubber.

3. Remove Water Droplets

Coalescer-Demister



4. Heat The Gas

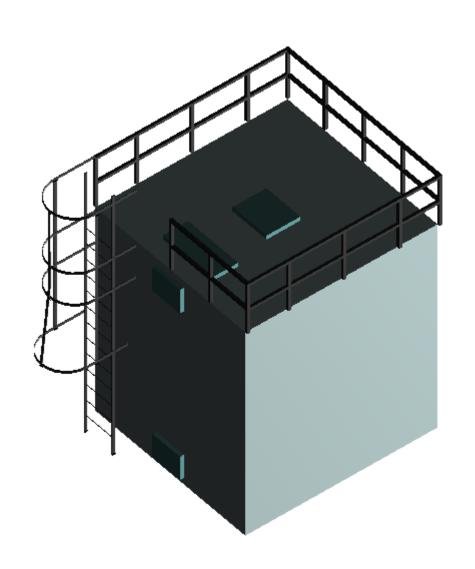


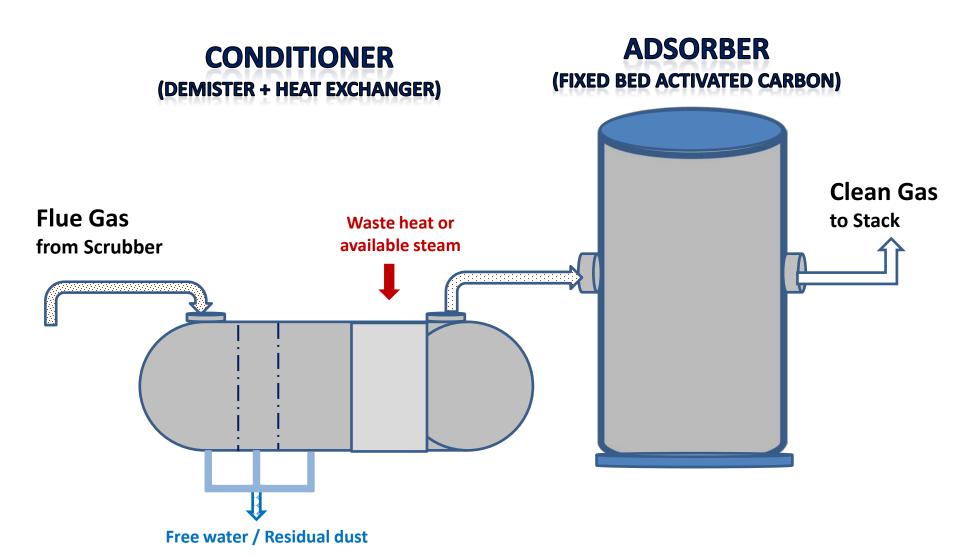
5. Remove Ultra-Fine Particles

Ultra-High Efficiency Filter



6. Adsorb the Hg





Pretreatment In One Container

